

MSE200

Lecture 18 (CH. 10.9, 10.11)

Polymeric Materials

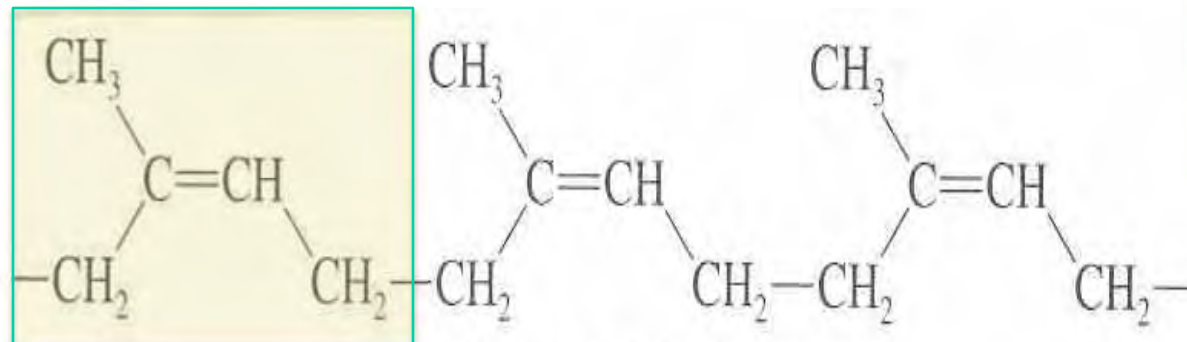
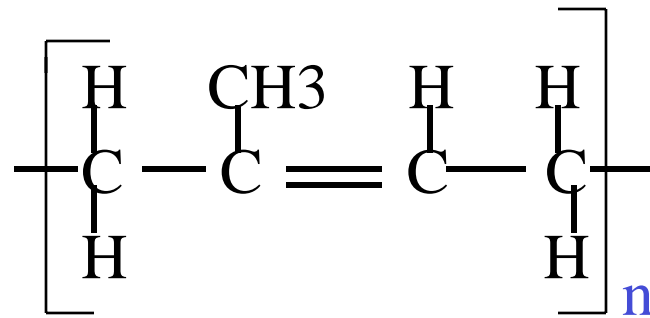
Instructor: Yuntian Zhu

Objectives/outcomes: You will learn the following:

- Elastomers and vulcanization.
- Mechanical properties of polymeric materials in relation to structure
- Stress relaxation

Elastomers (Rubbers)

- **Natural rubber:** Produced from latex (sap) of *Hevea Brasiliensis* tree

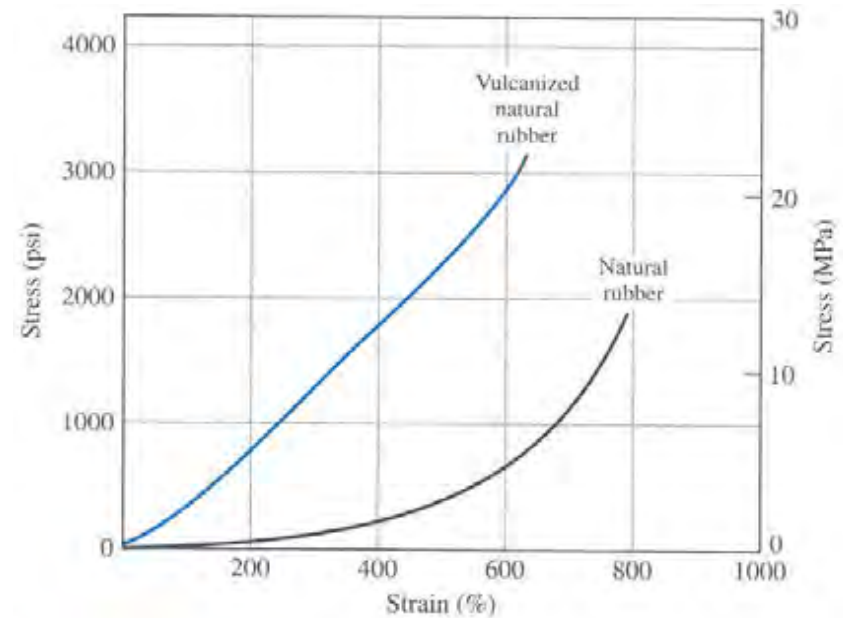
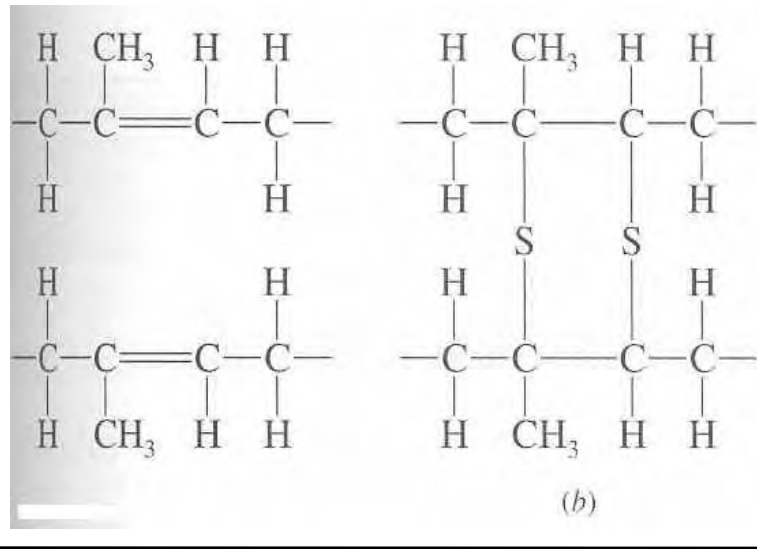


Segment of natural rubber polymer chain



Vulcanization:

- Heating rubber with sulfur and lead carbonate
 - Increases tensile strength
 - Increase stiffness
 - Restricts molecular movement by **crosslinking** of molecules

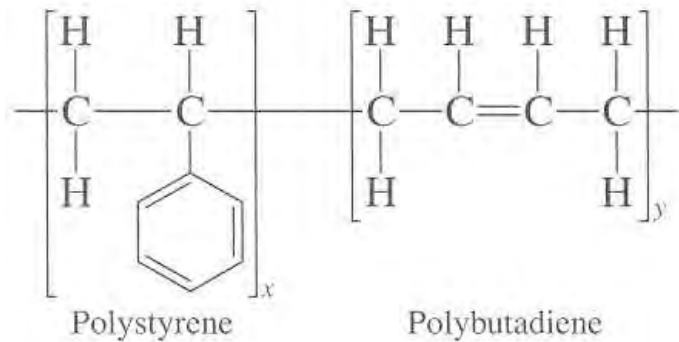


Natural and Synthetic Rubbers - Properties

Elastomer	Tensile strength (ksi) [†]	Elongation (%)	Density (g/cm ³)	Recommended operating temp.	
				°F	°C
Natural rubber* (<i>cis</i> -polyisoprene)	2.5–3.5	750–850	0.93	–60 to 180	–50 to 82
SBR or Buna S* (butadiene-styrene)	0.2–3.5	400–600	0.94	–60 to 180	–50 to 82
Nitrile or Buna N* (butadiene-acrylonitrile)	0.5–0.9	450–700	1.0	–60 to 250	–50 to 120
Neoprene* (polychloroprene)	3.0–4.0	800–900	1.25	–40 to 240	–40 to 115
Silicone (polysiloxane)	0.6–1.3	100–500	1.1–1.6	–178 to 600	–115 to 315

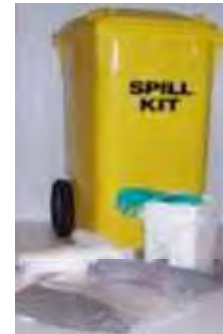
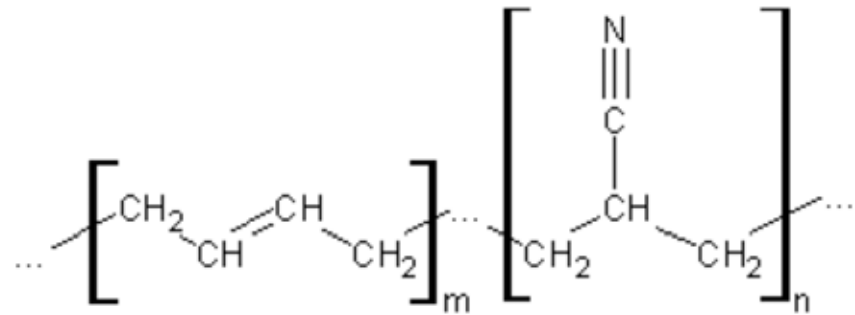
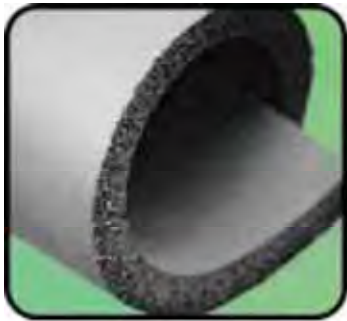
Synthetic Rubbers

- **Styrene-Butadiene rubber (SBR):** Most widely used.
 - Greater elasticity than natural rubbers.
 - Tougher and stronger, wear resistant.
 - Absorbs organic solvents and swell.



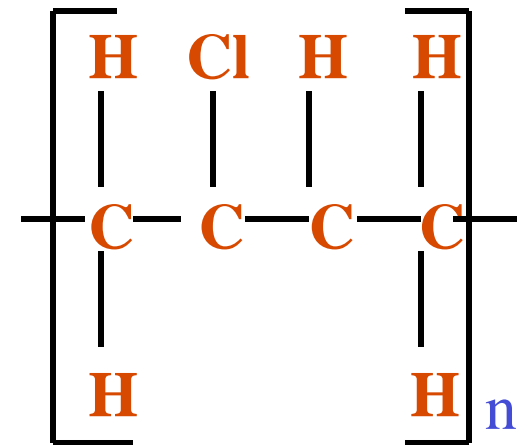
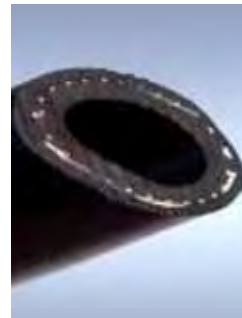
Nitrile Rubbers

- **55-82% Butadiene and 45-18% acrylonitrile.**
 - Resistance to solvents and wear
 - Less flexible.



Polychloroprene

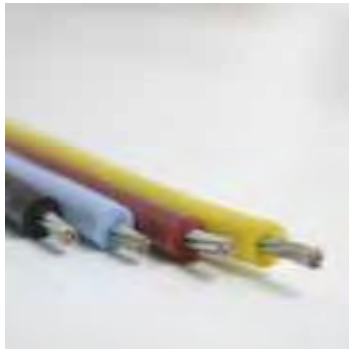
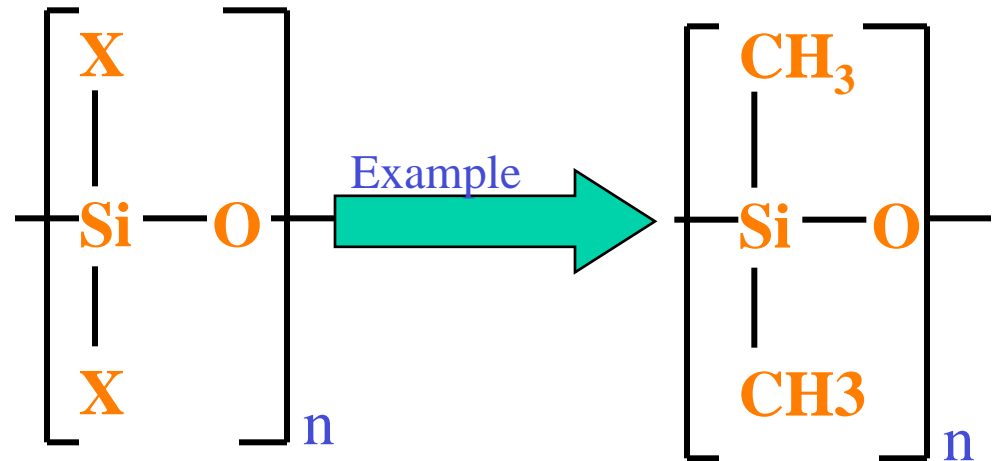
- Increased resistance to oxygen, ozone, heat and weather.
- Low temperature flexibility, high cost.



Silicone Rubbers:

Wide temperature range.

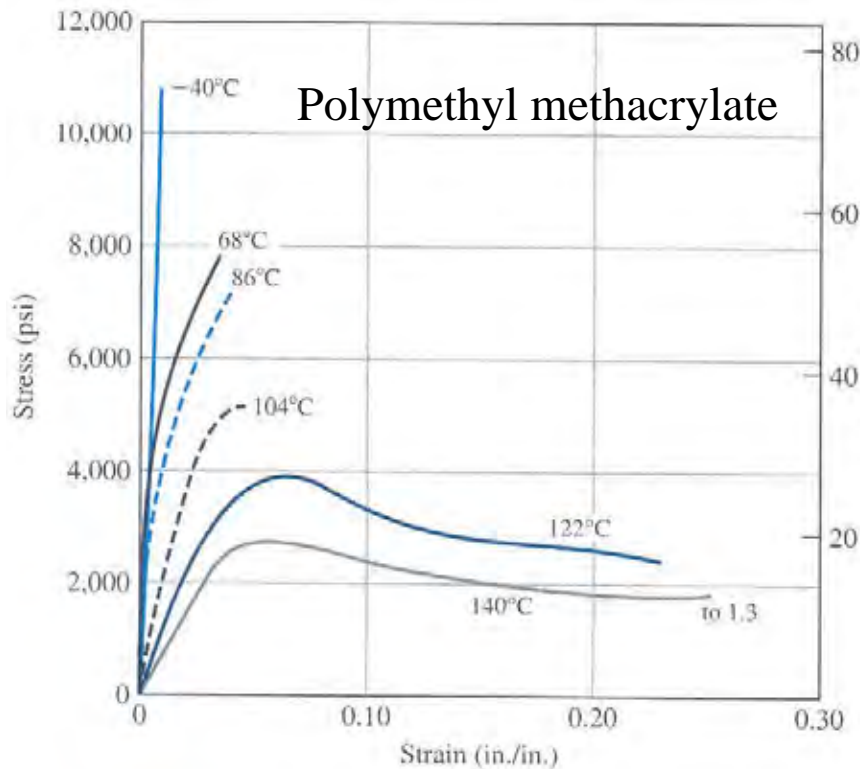
Used in gaskets, electric insulation etc.



<http://video.google.com/videosearch?q=silicone+rubber&hl=en&sitesearch=>

Deformation of Thermoplastics

- Below T_g : Elastic deformation.
- Above T_g : Plastic deformation.



Deformation mechanisms

Elastic deformation



Elastic or plastic deformation

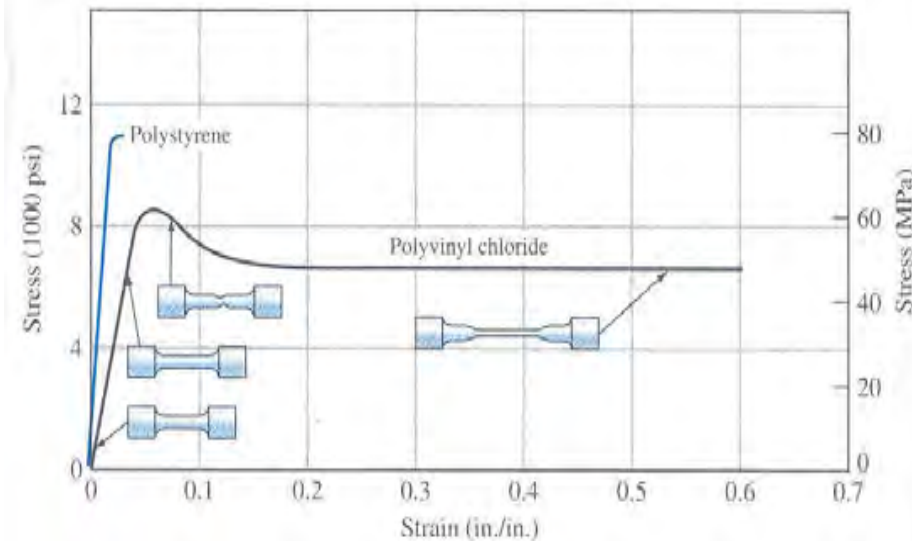


Plastic deformation



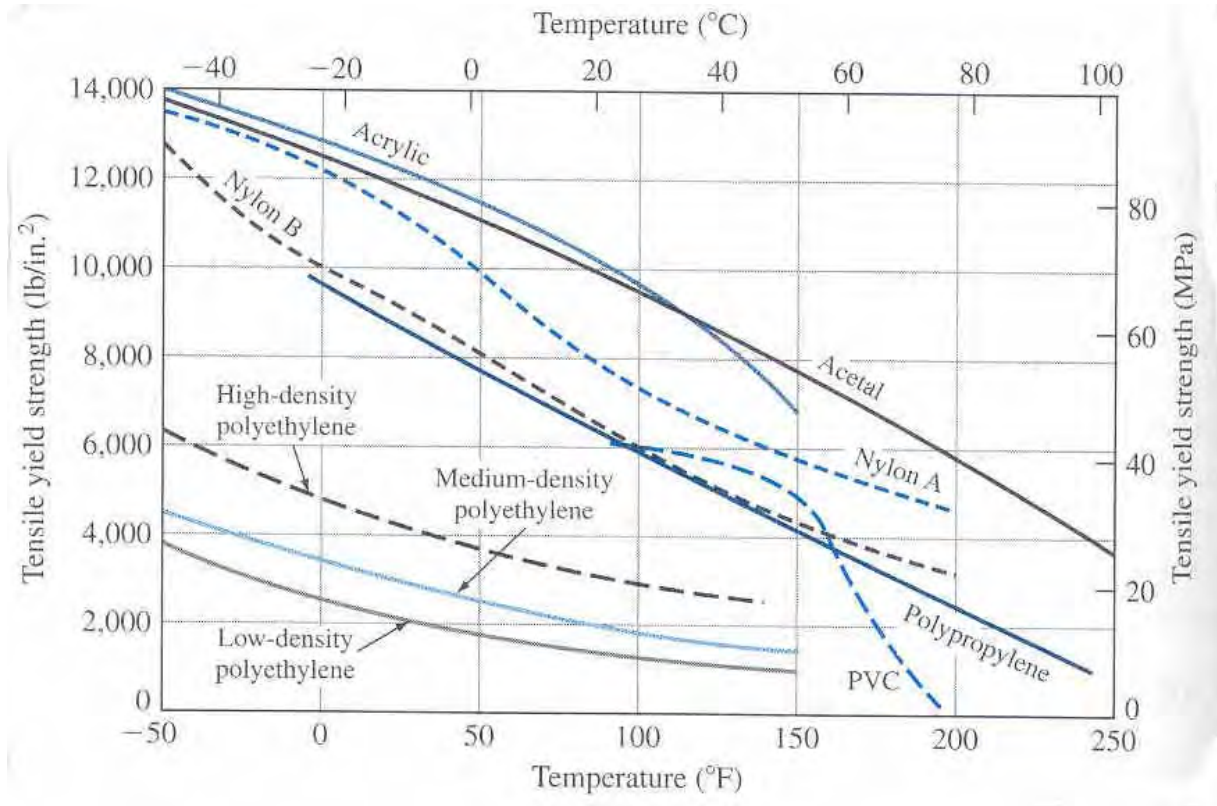
Strengthening of Thermoplastics

- Increasing **average molecular mass** increases strength up to a certain critical mass.
- Increasing the degree of **crystallinity** increases strength, modulus of elasticity and density.
- Introducing **pendant atomic groups** to hinder chain slippage during permanent deformation
- **Making composites by adding fibers/particles**



Effects of Temperature on Strength

- Thermoplastics
- Thermosets



Creep and Stress Relaxation of Polymers

- Creep increases with increased tensile stress and temperature.
- Stress relaxation: Decrease in stress at constant strain.

$$\sigma = \sigma_0 e^{\frac{-t}{\tau}}$$

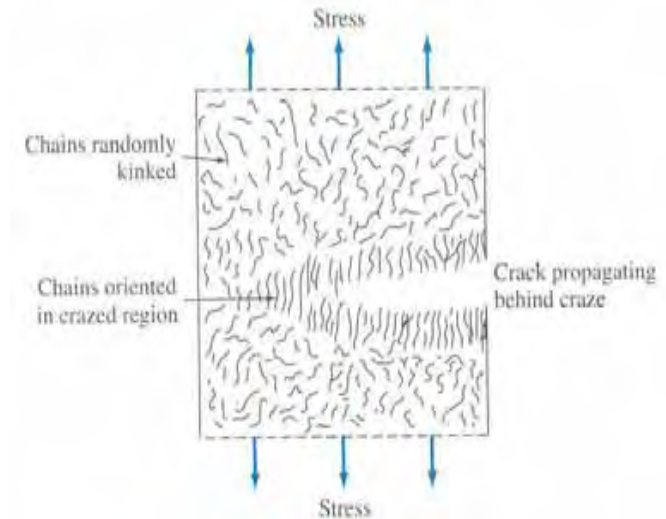
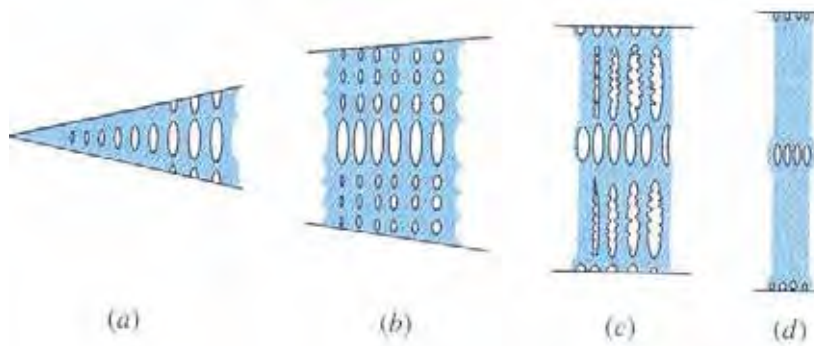
σ = Stress after time t.
 σ_0 = Initial stress

$$\frac{1}{\tau} = C e^{\frac{-Q}{RT}}$$

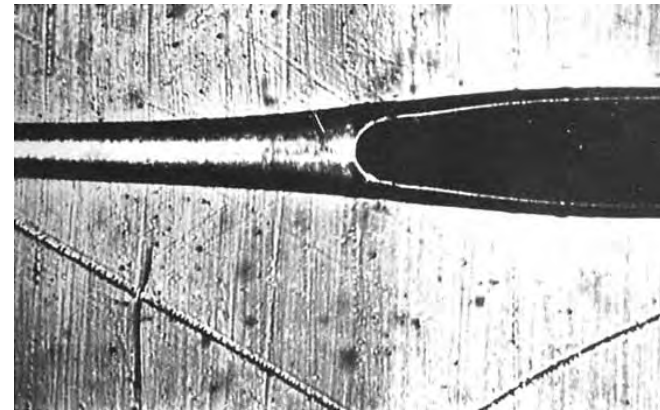
τ = relaxation time.
T = temperature, R = molar gas constant.

Fracture of Polymers

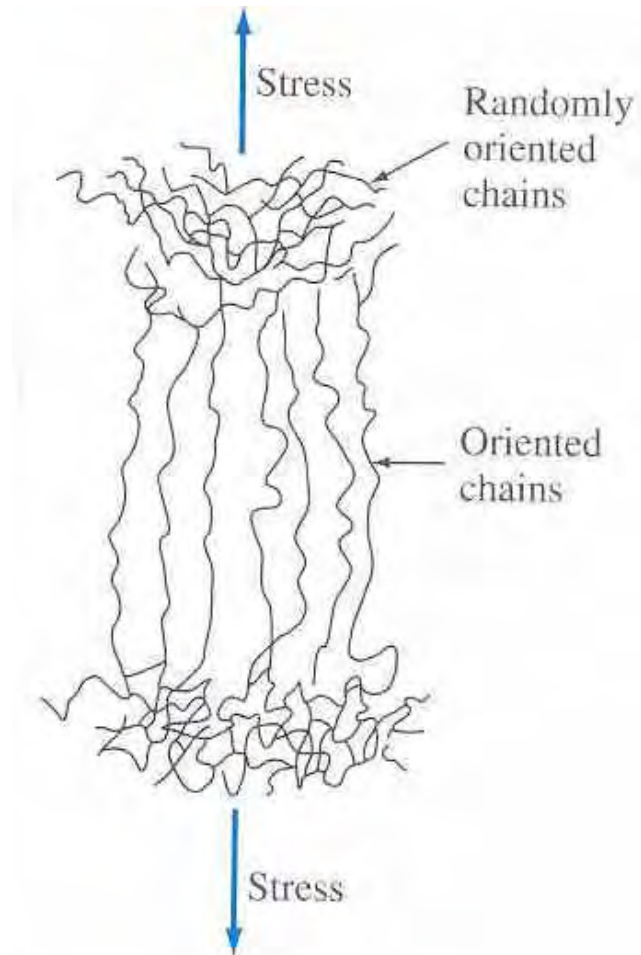
- **Thermosetting plastics: Primarily brittle mode.**
- **Thermoplastics: ductile or brittle depending on the temperature.**



Craze microstructure in a glassy thermoplastic



Ductile Fracture of polymer



HW

Chapter 10: 9, 13, 14, 20,

Example problems: 10.1, 10.2, 10.3, 10.4, 10.5, 10.6,
10.7, 10.8, 10.9, 10.11

Reading assignment: 11.1-11.3